R15

Code No: 121AE

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD B.Tech I Year Examinations, May/June - 2019 ENGINEERING CHEMISTRY

(Common to CE, EEE, ME, ECE, CSE, EIE, IT, MCT, AE, MIE, PTM, CEE)

Time: 3 hours Max. Marks: 75

Note: This question paper contains two parts A and B.

Part A is compulsory which carries 25 marks. Answer all questions in Part A. Part B consists of 5 Units. Answer any one full question from each unit. Each question carries 10 marks and may have a, b, c as sub questions.

PART- A

PART- A			
		(25 Marks)	
1.a)	What is the difference between chemical and electrochemical corrosion?	[2]	
b)	Write anode and cathode reactions of Ni-Cd cell.	[3]	
c)	What is a plastic? Give two examples.	[2]	
ď)	Give the applications of refractories.	[3]	
e)	What is hardness of water? Give its units.	[2]	
f)	What is Caustic embrittlement? What are its effects?	[3]	
g)	What is CNG? Give its composition and uses.	[2]	
h)	What are the advantages and disadvantages of solid fuels?	[3]	
i)	Define Phase and degree of freedom of a system.	[2]	
j)	Calculate the number of components in the following reaction	[3]	
3/	$NH_4Cl_{(s)} \Leftrightarrow NH_{3(g)} + HCl_{(g)}$		
	PART-B		
	Jen X	(50 Marks)	
	1031	,	
2.a)	Explain the construction and working principle of Quinhydrone electrode.		
b)	Explain the functioning of an electrochemical cell with example.	[5+5]	
	OR		
3.a)	Describe electroless plating of Nickel.		
b)	What is Cathodic protection? Explain sacrificial anode method.	[5+5]	
4.a)	Give the differences between thermoplastic and thermosetting resins with ex-	amples.	
b)	Write a note on conducting polymers.	[5+5]	
	OR		
5.a)	What do you mean by setting and hardening of cement? Discuss the v	arious steps	
	involved with the help of chemical equations.	_	
b)	Explain about cloud point, pour point and flash point of a lubricant.	[5+5]	
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6.a)	Explain Zeolite process of softening water.		
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7.a) Write a note on Boiler corrosion.

b)

b) Explain EDTA method of estimation of temporary and permanent hardness of water.

OR

What is reverse osmosis? Explain. Give the advantages of reverse osmosis.

[5+5]

[5+5]

8.a)	Give the criteria for selecting good fuel.	
b)	Explain ultimate analysis of coal and give its significance.	[5+5]
	OR	
9.a)	Explain the synthesis of petrol by Fischer Tropsch's process.	
b)	Explain the fixed bed catalytic cracking process with diagram.	[5+5]
,	Explain phase diagram of water system. Derive Langmuir adsorption isotherm.	[5+5]
	OR	
11.	Define the following terms: a) Annealing b) Hardening c) Normalization d) Chemisorption e) Isotherm	[10]

